

Effect of phase transfer chemistry, segmented fluid flow and sonication on the synthesis of cinnamic esters.



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Introduction and Scope of the Project

Phase-Transfer Catalyst (PTC) technology is used in the commercial manufacture of more than \$10 billion per year of chemicals, including pharmaceuticals, flavours & fragrances, dyes, additives and polymers.

PTC PROS¹

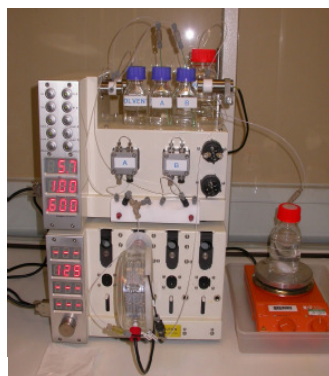
- ✓ Improved yield and selectivity
- ✓ Reduced variability
- ✓ Improved Environmental Performance
 - lower organic solvent consumption
 - less by-products
- ✓ Enhanced Safety
 - better control of exothermic reactions
 - use less hazardous reagents
- ✓ Reduced Manufacturing Costs
 - use alternate less expensive or easier to handle raw materials
 - avoid water sensitive reactants

PTC CONS¹

- ✗ Difficult scale up
- ✗ Long reaction time
- ✗ Mixing is crucial
- ✗ Presence of Phase transfer catalyst

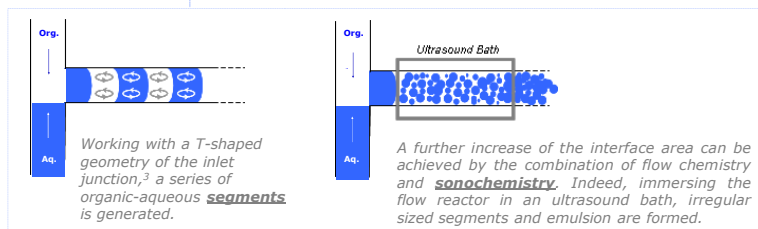
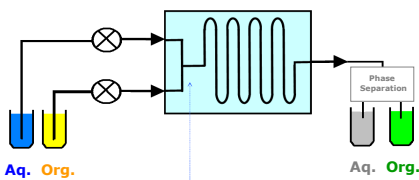
FLOW CHEMISTRY PROS²

- **Easy scale up**
- **Short reaction time**
- **High surface-to-volume ratio**
- Precise control of variables (e.g. temperature, pressure, concentration, residence time and heat transfer)
 - Superheating
 - Enhanced Safety

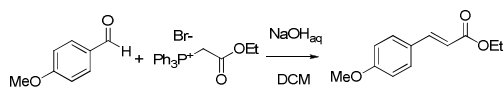


Vapourtec R2-R4 series Flow Reactor

Generic scheme of a bi-phasic reaction under flow chemistry conditions.



Model Reaction and Set-up of optimal conditions



System	Isolated Yield (%)	Reaction time
Flask reaction ⁴	55	24 h
Flow system	78	30 min ^a (RT ^b : 5 min)
Flow system without PTC	63	30 min (RT: 5 min)
Flow system without PTC and with sonication	96	30 min (RT: 5 min)

OPTIMAL CONDITIONS:

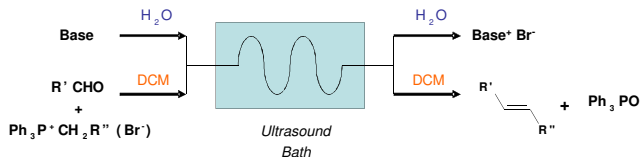
Reactor volume: **10 ml**
 Reagents Flow rate: **1 ml/min**
 Residence Time (RT): **5 min**
 Temperature: **RT**
 Organic solvent: **DCM**
 Stoichiometry:

- 1 eq.** aldehyde
- 2 eq.** phosphonium salt
- 2 eq.** base

- **PTC-free**
- **Only E-isomer achieved**

^a The reaction was performed on the same amount of starting material (1 mmol) as in the flask reaction system

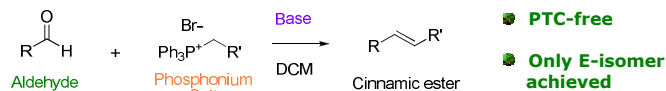
^b RT (retention time) = Reactor Volume (ml)/Tot Flow Rate (ml/min)



References

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Preparation of a set of cinnamic derivatives



Entry ^a	Aldehyde	Phosphonium Salt	Base	Isolated Yield ^{b,c} (%)
1			NaOH	96
2			NaOH	98
3			NaOH	98
4			NaOH	83 ^d
5			NaOH	86
6			NaOH	37
7			NaOH	63
8			NaOH	53
9			NaOH	75
10			NaOH	100
11			NaOH	48
12			LiOH	-
13			K ₂ CO ₃	98
14			NaHCO ₃	78
15			DMAP	78
16			NaOH	95
17			NaOH	42 ^d
18			NaOH	48 ^e

Aldehyde investigation

Base investigation

Phosphonium salt investigation

^a Reactor volume: 10 mL; Reagents Flow rate: 1 mL min⁻¹; Temperature: room temperature; Organic solvent: DCM; Molar ratio: aldehyde/ phosphonium salt/ Base=1/2/2 (without PTC).

^b The Z-stereoisomer was not revealed in UPLC-MS analysis.

^c Product purified by flash chromatography on silica gel.

^d E:Z stereoisomers in 6:1 ratio, as determined by ¹H-NMR.

^e E:Z stereoisomers in 1.4:1 ratio, as determined by ¹H-NMR.

Conclusions

- It has been proved that combination of phase transfer chemistry, flow chemistry and sonochemistry can lead to significant improvements in bi-phasic reactions.
- A simple, mild and fast protocol for the synthesis of cinnamic ester derivatives has been developed. This protocol:

- can be **applied** to different aldehydes, alkyl phosphonium salts and bases
- avoids phase transfer **catalysts**
- allows short **reaction time** (5 minutes residence time)
- leads to good **yield** and high **E/Z-selectivity**
- allows an easy **scale up**